

A Redetermination of the Barrier to Internal Rotation of 2-Fluoropropene

G. Bestmann and H. Dreizler
Abteilung Chemische Physik im Institut für Physikalische Chemie der Universität Kiel

Z. Naturforsch. 40 a, 267–270 (1985); received January 15, 1985

We present the determination of the methyl barrier to internal rotation from the rotational spectrum in the ground state of 2-fluoropropene. A complete set of barrier parameters could be determined. The results are compared with those of 2-chloro- and 2-bromopropene, achieved by the same technique and evaluation method.

Recently we determined the barrier to internal rotation of the methyl group from rotational transitions in the torsional ground state of 2-chloro- and 2-bromopropene [1, 2]. Now we present a redetermination of the barrier to internal rotation of 2-fluoropropene. The measurements were made by microwave Fourier transform spectroscopy [3–6] because of the higher resolution compared to Stark spectroscopy.

2-fluoropropene, CH3–CF=CH2, was first investigated by Pierce and O'Reilly [7]. The substance was purchased from PCR Research Chemicals, Gainesville, Florida, USA, and used after vacuum distillation. The spectra between 8 and 26 GHz were recorded in the pressure range between 1 to 2 mTorr and at a temperature of approximately –30 °C. Our measurements are given in Table 1. We performed a sixth order centrifugal distortion analysis of 83 A-species lines with a standard deviation of 5 kHz. The Hamiltonian of van Eijck [8] and Typke [9] (Eq. (6)) was used. It is worth to mention, that for 2-bromopropene only one centrifugal distortion parameter δJ could be determined. In 2-chloropropene a fourth order analysis was possible. The results for 2-fluoropropene are given in Table 2.

The barrier V3 was determined from 77 A–E splittings given in Table 1. The closely split doublets were corrected for effects of overlapping by a line shape analysis [10]. The internal axis method (IAM) was applied with a version of Wood's program [11–14]. Using the rotational constants of Table 2 a fit of the Fourier coefficient w1(s), s the reduced

Table 1. Measured transitions of 2-fluoropropene CH3–CF=CH2. Γ: symmetry species, νm: measured frequency, deviation from calculated frequency in brackets in units of the last digit, Δνm: measured internal rotation splitting νA–νE, Δνc: calculated internal rotation splitting, all frequencies in [MHz]. (****: not resolved).

Table with 10 columns: J, K-, J', K', K+, Γ, νm, Δνm, Δνc. It contains 24 rows of spectroscopic data for 2-fluoropropene.

Reprint requests to Prof. Dr. H. Dreizler, Institut für Physikalische Chemie der Universität Kiel, Olshausenstr. 40, D-2300 Kiel.

0340-4811 / 85 / 0300-0267 \$ 01.30/0. – Please order a reprint rather than making your own copy.



Dieses Werk wurde im Jahr 2013 vom Verlag Zeitschrift für Naturforschung in Zusammenarbeit mit der Max-Planck-Gesellschaft zur Förderung der Wissenschaften e.V. digitalisiert und unter folgender Lizenz veröffentlicht: Creative Commons Namensnennung-Keine Bearbeitung 3.0 Deutschland Lizenz.
Zum 01.01.2015 ist eine Anpassung der Lizenzbedingungen (Entfall der Creative Commons Lizenzbedingung „Keine Bearbeitung“) beabsichtigt, um eine Nachnutzung auch im Rahmen zukünftiger wissenschaftlicher Nutzungsformen zu ermöglichen.

This work has been digitalized and published in 2013 by Verlag Zeitschrift für Naturforschung in cooperation with the Max Planck Society for the Advancement of Science under a Creative Commons Attribution-NoDerivs 3.0 Germany License.
On 01.01.2015 it is planned to change the License Conditions (the removal of the Creative Commons License condition "no derivative works"). This is to allow reuse in the area of future scientific usage.

Table 1 (continued)

J	K_-	K_+	J'	K'_-	K'_+	Γ	v_m	Δv_m	Δv_c
3	3	1	3	2	2	A E	17536.043 (-6)	0.228	0.216
4	3	1	4	3	2	A E	8855.279 (6)	-0.125	-0.122
4	2	2	4	2	3	A E	21323.134 (2)	-0.022	-0.024
4	4	1	4	2	2	A E	22664.532 (7)	0.505	0.500
4	3	2	4	1	3	A E	24168.354 (-1)	0.178	0.179
4	3	1	4	2	2	A E	11542.545 (-11)	0.069	0.064
4	4	0	4	3	1	A E	12389.542 (0)	0.381	0.382
4	2	2	4	1	3	A E	21481.104 (-1)	*****	0.006
4	4	1	4	3	2	A E	19977.286 (4)	0.315	0.314
4	3	2	4	2	3	A E	24010.385 (1)	0.158	0.161
5	3	2	5	3	3	A E	19369.057 (-1)	-0.132	-0.129
5	4	2	5	2	3	A E	25551.627 (4)	0.283	0.289
5	5	0	5	4	1	A E	16984.808 (0)	0.559	0.558
5	3	2	5	2	3	A E	19977.043 (1)	-0.073	-0.063
5	4	1	5	3	2	A E	12291.698 (-5)	0.181	0.177
5	4	2	5	3	3	A E	24943.638 (6)	0.212	0.222
5	5	1	5	4	2	A E	23028.402 (3)	0.429	0.434
6	4	2	6	4	3	A E	16777.974 (4)	-0.250	-0.241
6	5	1	6	4	2	A E	14285.794 (1)	0.352	0.357
6	6	0	6	5	1	A E	22315.149 (-8)	0.714	0.708
6	4	2	6	3	3	A E	18479.540 (0)	-0.066	-0.066
6	5	2	6	4	3	A E	26421.425 (3)	0.303	0.312
7	5	2	7	5	3	A E	13753.051 (-7)	-0.331	-0.340
7	5	2	7	4	3	A E	17560.317 (2)	0.014	0.016
7	6	1	7	5	2	A E	17668.868 (-1)	0.591	0.588
8	5	3	8	5	4	A E	26309.600 (-3)	-0.316	-0.325
8	6	2	8	6	3	A E	10556.222 (-2)	-0.406	-0.405
8	7	1	8	6	2	A E	22348.635 (-3)	0.827	0.827
8	6	2	8	5	3	A E	17726.555 (0)	0.201	0.191
9	6	3	9	6	4	A E	22866.558 (-1)	-0.487	-0.490
9	7	2	9	6	3	A E	19319.144 (2)	0.445	0.447
9	6	3	9	5	4	A E	25116.583 (-3)	-0.196	-0.204

Table 1 (continued)

J	K_-	K_+	J'	K'_-	K'_+	Γ	v_m	Δv_m	Δv_c
10	7	3	10	7	4	A E	18900.165 (4)	-0.619	-0.621
10	7	3	10	6	4	A E	23523.886 (4)	-0.099	-0.090
10	8	2	10	7	3	A E	22497.024 (4)	0.754	0.757
11	8	3	11	8	4	A E	14732.496 (0)	-0.695	-0.694
11	8	3	11	7	4	A E	23072.795 (-8)	0.154	0.146
12	9	3	12	9	4	A E	10727.492 (1)	-0.689	-0.689
12	9	3	12	8	4	A E	24182.945 (3)	0.477	0.482
13	9	4	13	9	5	A E	24322.914 (8)	-0.943	-0.935
14	10	4	14	10	5	A E	19253.228 (4)	-1.022	-1.016
15	11	4	15	11	5	A E	14325.285 (1)	-0.998	-0.995
16	12	4	16	12	5	A E	9950.055 (-2)	-0.873	-0.873
17	12	5	17	12	6	A E	24117.334 (-2)	-1.330	-1.339
18	13	5	18	13	6	A E	18294.926 (-4)	-1.296	-1.306
19	14	5	19	14	6	A E	13019.608 (-1)	-1.153	-1.151
20	15	5	20	15	6	A E	8668.186 (-4)	-0.898	-0.909
21	15	6	21	15	7	A E	22635.158 (1)	-1.597	-1.594
22	16	6	22	16	7	A E	16460.846 (0)	-1.419	-1.415
23	17	6	23	17	7	A E	11235.556 (-2)	-1.137	-1.133
25	18	7	25	18	8	A E	20278.513 (0)	-1.650	-1.642
26	19	7	26	19	8	A E	14157.112 (2)	-1.310	-1.332
27	20	7	27	20	8	A E	9307.388 (-2)	-0.977	-0.976
28	20	8	28	20	9	A E	24475.415 (-6)	-1.812	-1.814
29	21	8	29	21	9	A E	17444.901 (15)	-1.510	-1.489
30	22	8	30	22	9	A E	11726.740 (0)	-1.108	-1.108
32	23	9	32	23	10	A E	21109.122 (1)	-1.589	-1.593
33	24	9	33	24	10	A E	14487.864 (3)	-1.204	-1.202
34	25	9	34	25	10	A E	9419.103 (1)	-0.810	-0.822
35	25	10	35	25	11	A E	25157.677 (7)	-1.639	-1.639
36	26	10	36	26	11	A E	17607.977 (7)	-1.253	-1.251
37	27	10	37	27	11	A E	11678.252 (7)	-0.871	-0.867
39	28	11	39	28	12	A E	21102.458 (11)	-1.252	-1.257
40	29	11	40	29	12	A E	14265.674 (9)	-0.886	-0.881

Table 2. Rotational and centrifugal distortion constants of 2-fluoropropene $\text{CH}_3\text{-CF=CH}_2$. σ : standard deviation of the fit, standard errors in units of the last digit in brackets.

$A = 10169.70848(108)$ MHz	$D'_J = 2.440(89)$ kHz	$H'_J = 0.000$ fixed
$B = 9041.63008(109)$ MHz	$D'_{JK} = 12.997(13)$ kHz	$H'_{JK} = -3.80(619)$ Hz
$C = 4927.06935(109)$ MHz	$D'_K = -4.813(9)$ kHz	$H'_{KJ} = 4.37(140)$ Hz
	$\delta'_J = 1.643(12)$ kHz	$H'_K = -2.37(72)$ Hz
	$R'_6 = -0.830(6)$ kHz	$H'_5 = -0.79(33)$ Hz
		$H'_6 = 3.88(145)$ Hz
	$\sigma = 5$ kHz	$H'_{10} = -1.98(70)$ Hz

Correlation coefficients

1.00																			
0.99	1.00																		
0.91	0.92	1.00																	
0.85	0.86	0.89	1.00																
0.11	0.17	-0.09	0.04	1.00															
0.08	0.02	-0.06	-0.05	-0.46	1.00														
0.13	0.18	-0.10	0.05	0.89	-0.07	1.00													
0.10	0.15	-0.07	0.07	0.88	-0.16	0.98	1.00												
-0.08	-0.02	0.06	0.09	0.42	-0.48	0.40	0.56	1.00											
0.08	0.01	-0.06	-0.08	-0.38	0.44	-0.37	-0.53	-0.99	1.00										
-0.08	0.01	0.05	0.07	0.32	-0.32	0.35	0.49	0.92	-0.97	1.00									
-0.03	0.01	0.04	0.09	0.51	-0.35	0.56	0.70	0.95	-0.91	0.82	1.00								
0.06	0.01	-0.06	-0.09	-0.44	0.45	-0.45	-0.60	-0.99	0.97	-0.90	-0.98	1.00							
-0.09	-0.02	0.07	0.08	0.38	-0.51	0.34	0.50	0.99	-0.99	0.93	0.92	-0.98	1.00						

Table 3. Internal rotation parameters of 2-fluoropropene, $\text{CH}_3\text{-CF=CH}_2$. $w_1(s)$: Fourier coefficient, $\angle(a, i)$: angle between the inertia axis a and the internal rotation axis i , I_x : moment of inertia of the methyl group, $(\cdots)$: correlation coefficient, σ : standard deviation of the fit, Δv : mean experimental splitting, s : reduced barrier height, F : reduced rotational constant of the internal rotation, V_3 : barrier to internal rotation, N : Number of splittings, standard errors in units of the last digit in brackets.

$w_1(s)$	$-0.9256(93) \cdot 10^{-5}$
$\angle(a, i) [^\circ]$	5.84(75)
$I_x [\text{amu}\text{\AA}^2]$	3.177(18)
$(w_1(s), \angle(a, i))$	-0.995
$(w_1(s), I_x)$	0.970
$(\angle(a, i), I_x)$	-0.980
$\sigma [\text{kHz}]$	7
$\Delta v [\text{kHz}]$	583
s	65.41(9)
$F [\text{GHz}]$	169.929(96)
$V_3 [\text{kcal/mol}]$	2.384(17)
N	77

barrier, $\angle(a, i)$, the angle between the inertia axis a and the internal rotation axis i and I_x , the moment of inertia of the methyl group, converged to reasonable values as given in Table 3. For this molecule the value of $I_x = 3.177(18) \text{ amu}\text{\AA}^2$ could be fitted, which is near to the value of $I_x = 3.167 \text{ amu}\text{\AA}^2$ for 2-chloropropene derived in [1] from a complete r_s -structure [15].

Table 5 of [2] may now be completed with measurements taken by the same technique and evaluated by the same method. The comparison is given in Table 4. The general trend in variation of the barrier height V_3 is like for $\text{CH}_3\text{CH}_2\text{X}$, $\text{X} = \text{F}$, Cl , and Br . V_3 increases from F to Cl and Br .

We thank the members of our group for help, the Deutsche Forschungsgemeinschaft and Fonds der Chemie for funds. The calculations were made at the computer center of the University of Kiel.

Table 4. Comparison of potential barriers V_3 and related parameters of 2-halopropenes $\text{CH}_3\text{-CX=CH}_2$, $\text{X} = \text{F}$, Cl and Br . Assumptions in square brackets.

X	F	^{35}Cl	^{37}Cl	^{79}Br	^{81}Br
$V_3 [\text{cal/mol}]$	2384(17)	2568(10)	2582(12)	2571(49)	2558(74)
$\angle(a, i) [^\circ]$	5.84(75)	59.78(84)	61.28(108)	56.8(34)	56.9(48)
$I_x [\text{amu}\text{\AA}^2]$	3.177(18)	[3.167]	[3.167]	[3.167]	[3.167]
$F [\text{GHz}]$	169.93(96)	165.91	165.69	164.64	164.72
s	65.41(9)	72.15(29)	72.63(35)	72.8(14)	72.4(21)
	this work	[1]	[1]	[2]	[2]

- [1] E. Fliege and H. Dreizler, *Z. Naturforsch.* **38a**, 1231 (1983).
- [2] E. Fliege and H. Dreizler, *Z. Naturforsch.* **39a**, 637 (1984).
- [3] G. Bestmann, H. Dreizler, H. Mäder, and U. Andresen, *Z. Naturforsch.* **35a**, 392 (1980).
- [4] G. Bestmann and H. Dreizler, *Z. Naturforsch.* **37a**, 58 (1982).
- [5] G. Bestmann, H. Dreizler, E. Fliege, and W. Stahl, *J. Mol. Struct.* **97**, 215 (1983).
- [6] W. Stahl, E. Fliege, H. Dreizler, and R. Schwarz, *Z. Naturforsch.* **39a**, 354 (1984), and to be published.
- [7] L. Pierce and J. M. O'Reilly, *J. Mol. Spectr.* **3**, 536 (1959).
- [8] B. P. van Eijck, *J. Mol. Spectr.* **53**, 246 (1974).
- [9] V. Typke, *J. Mol. Spectr.* **63**, 170 (1976); program ZFAP6 by V. Typke, 1982.
- [10] E. Fliege and H. Dreizler, *Z. Naturforsch.* **39a**, 630 (1984).
- [11] R. C. Woods, *J. Mol. Spectr.* **21**, 4 (1966).
- [12] R. C. Woods, *J. Mol. Spectr.* **22**, 49 (1967).
- [13] J. Meier, Dissertation 4611, ETH Zürich, 1970.
- [14] Modified by H. Lutz, E. Fliege, and G. Bestmann; programs IAMCAL, IAMFIT and IAMMOD.
- [15] W. Good, R. J. Cohan, jr., A. Bauder, and Hs. H. Günthard, *J. Mol. Spectr.* **41**, 381 (1972).